Accuracy Improvements in Service Life Testing Using FTIR Spectrometry

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What is FTIR?

Fourier Transform Infrared Spectrometry. *Fourier Transform:* Math used to convert Signal to a Spectrum *Infrared:* Wavelengths beyond visible light (where almost everything absorbs) *Spectrometry:* Using light absorption.

FTIR uses an "interferometer" -- a device which splits a beam of light, and the recombines it to produce an *Interferogram*. Software converts the Interferogram to an *Infrared Absorbance* spectrum.

A computer Model fits the absorption spectrum and generates Gas Concentration Readings.

Why FTIR is Better than IR

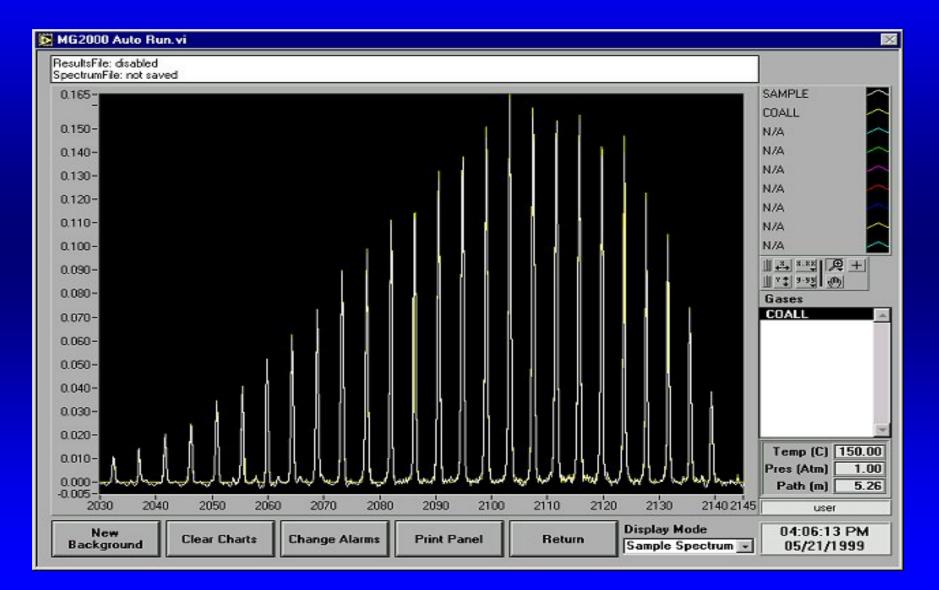
- Most chemicals have a distinct IR absorption spectrum
 With multiple IR Absorption Bands
- Absorption bands of different chemicals overlap
 - Water (H2O) has lots of bands that overlap
- FTIR makes IR absorption bands Narrower

 Improved Resolution
- FTIR absorption bands are more intense
 - Improved Sensitivity
- Chemo-metrics Software Curve Fitting Algorithm
 - Simultaneous measurement of 5 or more chemicals in a mixture

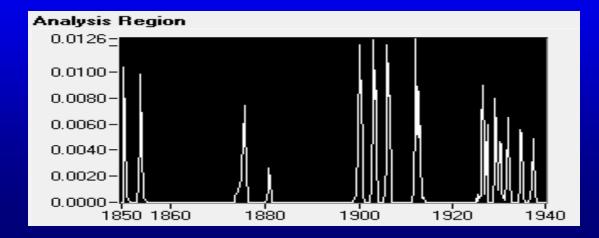
Features & Benefits of Detectors

Measurement of	EC	TLD	IR	FTIR
Different Gases & Vapors	1	12+	Many	Many
In presence of similar species	С	С	В	Α
In low ppm range	Yes	Yes	No	Yes
In high ppm to % Range	No	No	Yes	Yes
Continuously	Yes	Yes	Yes	Yes
Calibration Frequency	Daily	Rarely	Daily	Rarely
EC = Electrochemical Sensor	TLD = Colorimetric Tape Reader ChemCassette			

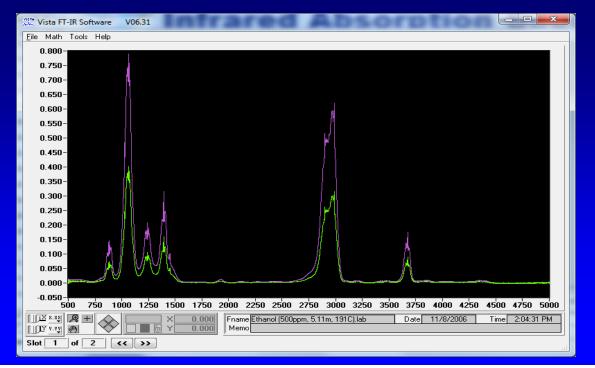
FTIR Spectrum of CO



FTIR Spectra

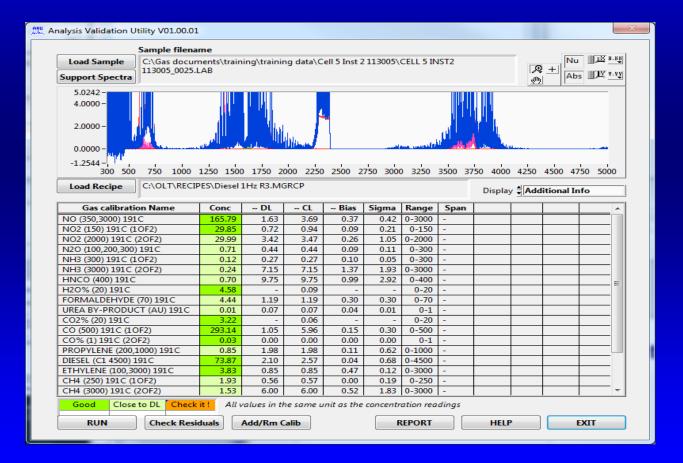


NO2



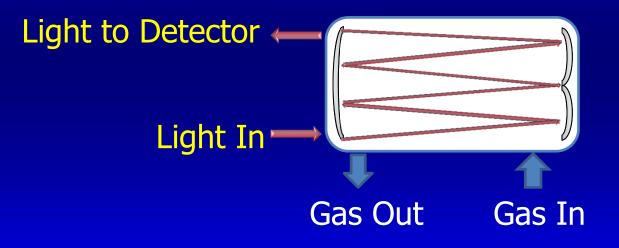
Ethanol

Single Method Includes Multiple Gas Calibrations



Gas Measuring Cell

Path Length = 5.11 Meter Cell Volume = 0.2 Liter



Cell Material = Dursan-coated SS

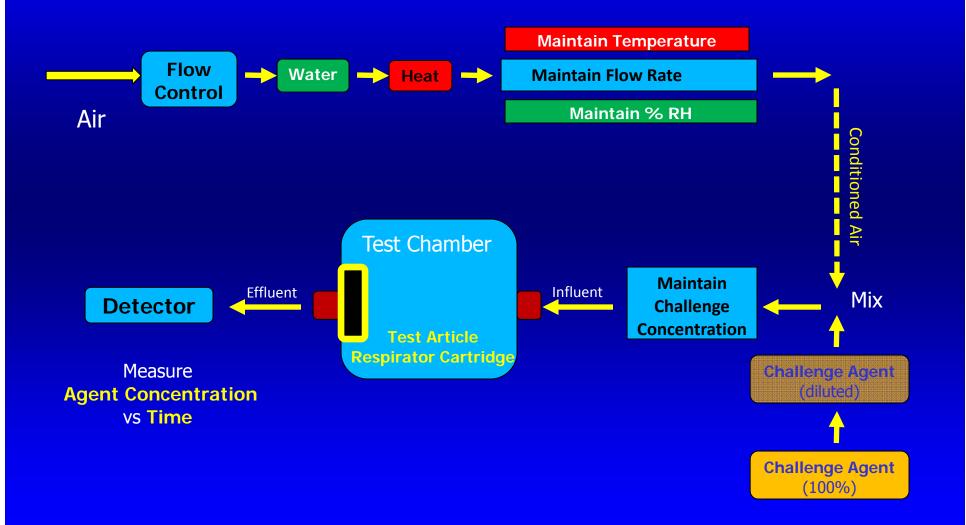
FTIR Spectrometer (front)



FTIR Spectrometer (rear)

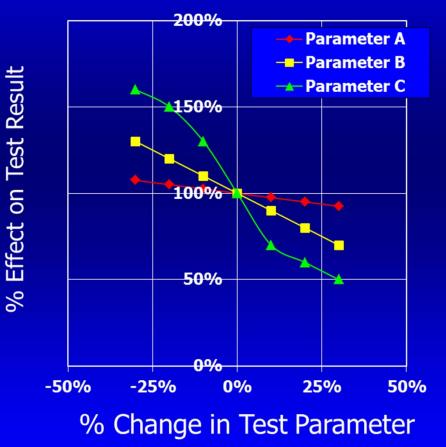


Anatomy of a Chemical Challenge Test



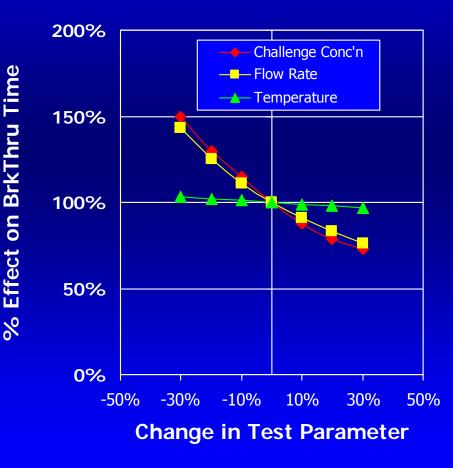
Sensitivity of Test Result to Test Parameters

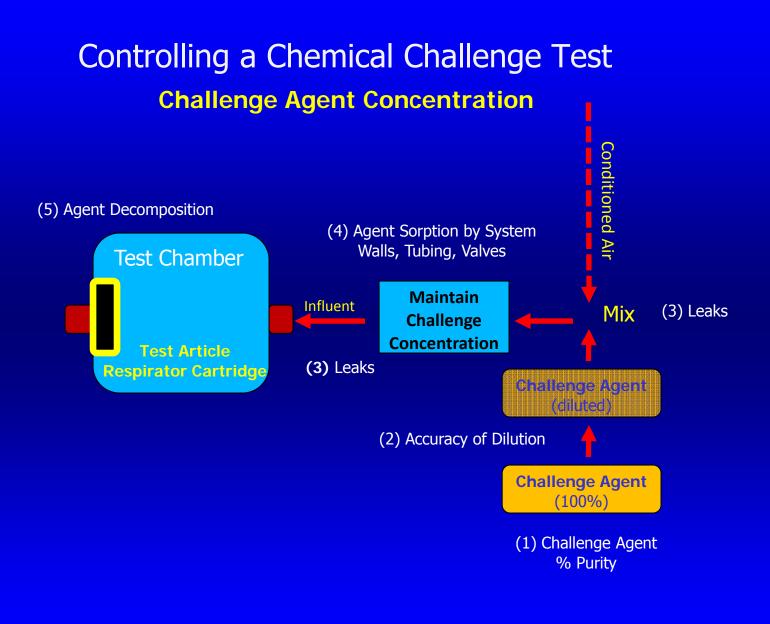
- The Test Result is more sensitive to variations in some Test Parameters than in others
 - A Not Very Sensitive
 - **B** Proportional
 - C Very Sensitive



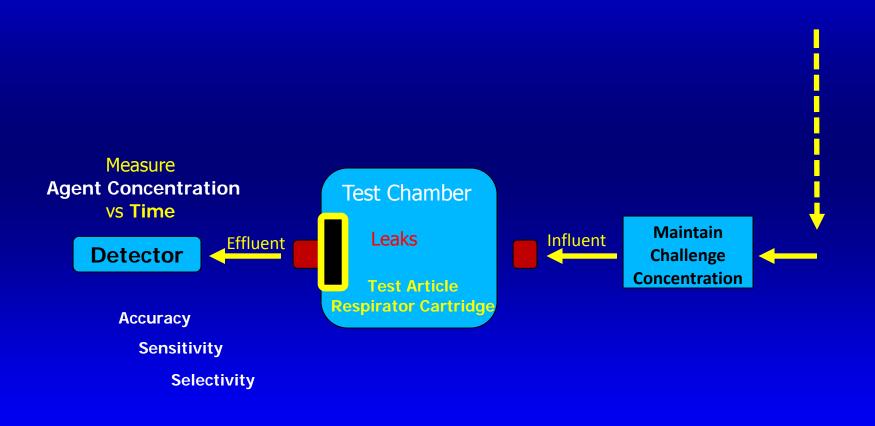
Sensitivity of Test Result to Challenge Agent Concentration

- The Test Result is more sensitive to some Test Parameters
 - Control of Challenge Agent Concentration
 - Control of Flow Rate
- Not so sensitive to
 - Temperature
 - (Measured as °C)





Controlling a Chemical Challenge Test Chamber and Detector



Control of Challenge Concentration

Test Parameter

- Challenge Agent Conc'n
 - Purchased Gas
 - Diluted gas
- Challenge Agent Conc'n
 - Loss by Decomposition
 - Loss by Sorption

Control Criteria

- Cylinder Gas Conc'n
 - Vendor Certificate
 - Wet Chem or FTIR Analysis
- Measure Gas *in situ*
 - Wet Chem Analysis
 - Prior to Test
 - FTIR
 - Continuous

Direct & Continuous Measurement of ...

Cylinder Gas

 As Received
 After Storage

Cylinder Gas Concentration Accuracy

 Control Accuracy of Challenge Level
 10% error leads to 10% error in BT Time

Direct & Continuous Measurement of ...

Influent Challenge Agent – Just Prior to Test

Challenge Gas Accuracy

- 10% error leads to 10% error in BT Time
- Errors Can Multiply
 - Cyl Gas Accuracy Error + Sorption Loss Error

Direct & Continuous Measurement of ...

- Effluent Agent - Breakthrough
- Agent Concentration Error
 25-50% error leads to 10% BT Time error

Agent Detection Error Insensitive Detection of Agent Misidentification of Agent

- Fail to Detect Reaction Products

Direct & Continuous Measurement of ...

Diluted Cylinder Gas

 As Received & Prior to Use

- Influent Challenge Agent

 Just Prior to Test
- Effluent Agent or Reaction Products

 Breakthrough

What Agents Can Be Analyzed by FTIR?

Direct & Continuous Measurement

In the Presence of Similar Agents

FTIR Calibrations (availability)

Now AVAILABLE

- FormaldehydeSulfur Dioxide
- CO & CO2
- Ammonia
- NO & NO2
- Chlorine Dioxide
- Carbon Tetrachloride

- Benzene
- HCl, HF, HBr
- CyclohexanePhosphine
- Phosgene
- Ozone
- Acrolein ... and others

To Be Developed

- a-chloroacetophenone (CN)
 o-chlorobenzylidene
- malononitrile (CS)
- Cyanogen Chloride

Can't Do

- C|2
- Br2
- **No Dipole Moment**

Summary of FTIR Applications

- Correct Challenge Agent Concentration
 - Unstable Cylinder Gas
 - Gas Sorption during delivery to test
 - Gas Decomposition during delivery
- Improve Breakthrough Detection
 - Agent Difficult to Detect
 - Detect Reaction Products
 - Detect Agents in presence of Reaction Products

Summary of Value (delivered by FTIR)

Improve Overall System Accuracy
 – Reduce Error

Improve Inter-Laboratory Test Agreement

 Improve overall Quality Assurance

AT Respirator and Filter Chemical Challenge Test Lab



